Nanostructure Evolution in AA7075 Alloy Produced by Solid State Additive Manufacturing – Additive Friction Stir - Deposition

Rekha M Y¹, Dustin Avery², Paul G Allison², J. Brian Jordon² and Luke Brewer³

¹Metallurgical and Materials Engineering, University of Alabama, Tuscaloosa, AL, USA, Tuscaloosa, Alabama, United States, ²Mechanical Engineering, University of Alabama, Tuscaloosa, AL, USA, Tuscaloosa, Alabama, United States, ³University of Alabama, Tuscaloosa, Alabama, United States

Solid-state additive manufacturing is a promising technology for producing complex geometry alloys for aerospace applications [1]. Among metal additive manufacturing technologies, additive friction stir-deposition (AFS-D) provides an innovative solution for the fabrication and repair of high strength Al-alloy components [2]. In the AFS-D process, feedstock material is deposited in a layer-by-layer fashion using a hollow, rotating tool onto a substrate [3]. When the hollow rotating tool comes in contact with the substrate, frictional heat and pressure is generated which creates a metallurgical bond between the feedstock material and the substrate. For the AA7075 alloy, the nanoscale precipitates MgZn₂ (η') and MgZn₂(η) control the strength of the material and may be dissolved and reformed by the thermal-mechanical AFS-D process. In this study, we investigate the evolution of size, phase constitution and spatial distribution of nanoscale precipitates in as-deposited and heat-treated AA7075 material.

Solid AA7075-T651 feedstock rod with a nominal composition of (wt.%) Al–5.6Zn–2.6Mg–1.6Cu was deposited onto a substrate (rolled plate with the same composition as the feedstock rod). Feedstock rod dimension was 9.5 mm (square geometry) while the damaged plate substrate with milled groove through the centre was 6.35 mm thick. Parameters used for the deposition were 225 RPM tool rotational speed, 50.8 mm/min traversing velocity and 50.8 mm/min feedstock feed rate. Sections of material produced by AFS-D were subsequently heat-treated to recover the T6 temper [4]. The heat-treatment schedule included solutionizing treatment at 490°C for 1 hr followed by water quench and then age hardening at 120°C for 24 hr.

Figures 1(a), (b) and (c) show the TEM bright field images of AFS-D, heat-treated AFS-D and heat affected zone (HAZ) AFS-D samples respectively. All the samples contained finely distributed Zn-Mg-Cu rich precipitates (dark precipitates in these images). Average size of the precipitates in as-deposited AFS-D, heat-treated AFS-D and HAZ AFS-D samples were 8.84 ± 2.8 nm, 7.18 ± 2.5 nm and 8.27 ± 2.9 nm respectively. The heat treated sample (Figure 1(b)) showed highest density of fine precipitates. The HAZ sample (Figure 1(c)) showed similar sized precipitates as AFS-D sample but with lower density. Few relatively larger sized precipitates were also be observed in HAZ sample due to the aging during the AFS-D process. The as-deposited AFS-D sample (Figure 1(a)) showed the finest precipitates, which were fewer in amount than those observed in the case of the heat treated sample. It is speculated that these precipitates formed due to natural aging of the sample after deposition. The SAED pattern from as-deposited material (Figure 1(d)) shows diffraction rings corresponding to randomly oriented (AlZn)Mg₈Zn₁₀ precipitates. The S-phase (Al₂CuMg), MgZn₂ (η) and MgZn₂ (η') phases are also present in this SAED pattern. In the heat affected zones caused by multiple passes of the tool (Figure 1(e)), the SAED pattern reveals the presence of MgZn₂ (η) and MgZn₂ (η') phases along with the presence of (AlZn)Mg₈Zn₁₀ phase. These diffraction patterns revealed that the heat-treated sample underwent in situ solutionization and re-precipitation of strengthening phases. For ex situ heat treated material, the SAED pattern (Figure 1(f)), reveals the presence of S-phase (Al₂CuMg), MgZn₂ (η) and (AlZn)Mg₈Zn₁₀ phases. With respect to the spatial location of the precipitates it should be further noted that the as-deposited AFS-D repair sample (Figure 2(a)) also contains large sized precipitates (127.94 ± 38.2 nm) which are discontinuously arranged over the grain-boundary region and within the grains in the microstructure. In contrast, the heat-treated AFS-D sample (Figure 2(b)) exhibited a distinctly different microstructure in which the fine sized precipitates (15.85 ± 5.1 nm) are distributed along the grain-boundaries, while the relatively
larger size precipitates (95.27 ± 25.9 nm) are distributed within the grains away from the grain-boundary region.

In the near future, atom probe tomography will be utilized to measure the chemical composition of these precipitates and to analyse the location and distribution of precipitate in as-deposited and heat-treated AA7075 alloys.

**Figure 1.** Representative TEM bright field images of (a) as-deposited AFS-D repair, (b) heat-treated AFS-D repair and (c) HAZ AFS-D. SAD patterns obtained from (d) as-deposited AFS-D repair, (e) heat-treated AFS-D repair and (f) HAZ AFS-D. The dotted lines in the SAD pattern are viewing aid.

**Figure 2.** Representative TEM bright field images of (a) as-deposited AFS-D repair and (b) heat-treated AFS-D repair samples.

References